SET-1

Subject Code: 22CH202BS

HT NO: 7 R

CMR TECHNICAL CAMPUS

UGC AUTONOMOUS

B. Tech. II Sem Supply End Examinations, January-2024 Engineering Chemistry Common to ECE, AIML, CSM&CSC

Time: 3 Hours

Max. Marks: 60

Note

- i. This Question paper contains Part- A and Part- B.
- ii. All the Questions in Part A are to be answered compulsorily.
- iii. All Questions from Part B are to be answered with internal choice among them.

PART-A

 $10 \times 01 = 10 \text{ Marks}$

		Marks	CO	BL
1. a	Define Break point Chlorination.	1	CO1	1
ь	List out the specifications of Potable water.	1	CO1	1
С	Write the Preparation of PVC.	1	CO2	1
d	Define Biodegradable polymer with an example.	1	CO2	1
e	What are the differences between Primary Cells and Secondary cells?	1	CO3	1
f	Define galvanization with an example.	1	CO3	1
g	Define cracking with an example.	1	CO4	1
h	Define Calorific value with units.	1	CO4	1
i	Write the structure of Polyvinyl amides.	1 *	CO5	1
j	Define the flash point and fire point of a lubricant.	1	CO ₅	1

PART- B

 $5 \times 10 = 50 \text{ Marks}$

			Marks	CO	BL
2.	a	Explain the estimation of the hardness of water by the EDTA method.	5	CO1	2
	b	Discuss desalination of water by Reverse osmosis method. OR	5	CO1	2
3	a	Define sludge and scale. Explain the formation of sludge and scale in the boilers.	3	CO1	2
	b	Explain the Ion exchange process for the purification of water.	7	CO1	2
4	a	Discuss the differences between thermoplastic and thermosetting plastics.	5	CO2	2
	b	Explain the vulcanization of natural rubber with advantages.	5	CO2	2

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	OR						
5	a		in the Preparation and engineering applications of BUNA-S.			CO2	20
	b Explain the mechanism of conduction for polyacetylene.			acetylene.	5	CO2	2
							1
6	a		the construction and working principle of a methanol			CO ₃	2
		lithium-ion battery.					_
	b					CO3	2
	oxygen fuel cell.						
_	OR					002	2
7	a	Explain the factors affect			5	CO3	2 2
	b	Discuss electroplating to	prevent the metal from	n corrosion.	5	CO3	2
0	_	Englain the Drawinsots of	malvois of anal and its	ionificance	5	CO4	2
8	a	Explain the Proximate at		significance.	5	CO4	2
	b Explain the process of Knocking. OR				3	CO4	2
9	a	Discuss the preparation	Sales S	Sischer	5	CO4	2
,	а	Tropsch's process.	ass the preparation of synthetic petrol by Fischer		3	COT	2
	b		e preparation of Biodiesel by Trans esterification		5	CO4	2
	and mention its advantages.				001		
			5-2-				
10	a	Explain Viscosity and V	ty and Viscosity index of a lubricant Discuss			CO5	2
		how Viscosity of a lubricant varies with temperature					
	b Summarize the Classification of lubricants.			5	CO5	2	
			OR				
11	a	Illustrate the mechanism	of thick film lubricati	on.	5	CO5	2
	b	Explain the setting and h	nardening of Portland of	ement.	5	CO5	2
СО	: 0	Course Outcomes					
BL	: E	Bloom's Taxonomy Levels	els L1: Remembering L2: U		derstandir	ng	
			L 3 : Applying	L 4 : An	alysing		

L 6 : Creating

L 5 : Evaluating